Initials:

- 1. (22 points)
- a. Rank fastest to slowest reaction with PhMgBr



b. Rank highest to lowest oxidation state



c. Provide pKa's for any 6 of the following compounds (if you do them all, we will count your best 6).



#### d. Fill in the correct nucleophile and electrophile from the table to complete the retrosyntheses.



2. (19 points) Fill in the boxes with the appropriate starting material, reagent or major product. Initials: Show stereochemistry where appropriate.



3. Provide an arrow-pushing mechanism (10 points).



Initials:

Is this reaction an oxidation, reduction, or neither?

Mechanism:			
p. 0	1. NaBH₄ 2. H <sub>3</sub> O⁺	→ OH	Is this reaction an oxidation, reduction, or neither?
Mechanism:			

Initials:

4. Propose syntheses of the targets below (14 points). **All carbons** must come from the starting materials provided, you can use any reagent you wish. **YOU CAN IGNORE STEREOCHEMISTRY.** 



Target A.



# Target B.



Initials:

# 5. Propose syntheses of the targets below (10 points).

All carbons must come from the starting materials provided, you can use any reagent you wish. YOU CAN IGNORE STEREOCHEMISTRY.



# Target A.



# Target B.

