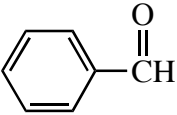
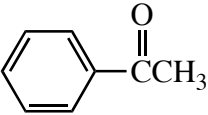
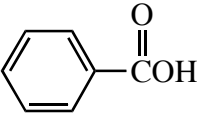
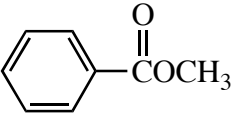


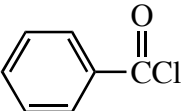
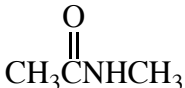
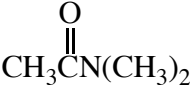
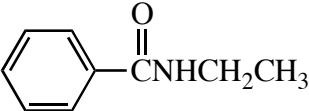


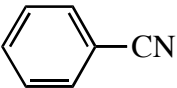
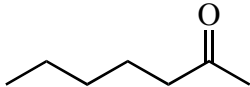
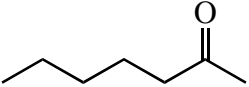
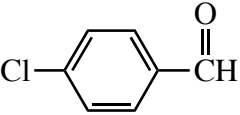
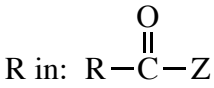


	Common <i>use common prefix name (see p. 2)</i>	IUPAC	Attached to a Phenyl Ring
Aldehyde	Add <i>-aldehyde</i> to common prefix name $\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_3\text{CH} \end{array}$	Change <i>-e</i> to <i>-al</i> in base name $\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_3\text{CH} \end{array}$	<i>-benzaldehyde</i> 
Ketone	Name 2 groups on carbonyl & add <i>-ketone</i> $\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_3\text{CCH}_2\text{CH}_3 \end{array}$	Change <i>-e</i> to <i>-one</i> in base name. Indicate location of carbonyl with a number. $\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_3\text{CCH}_2\text{CH}_3 \end{array}$	Add <i>-ophenone</i> to common prefix name 
Carboxylic Acid	Add <i>-ic acid</i> to common prefix name $\begin{array}{c} \text{O} \\ \parallel \\ \text{HCOH} \end{array}$	Change <i>-e</i> to <i>-oic acid</i> in base name $\begin{array}{c} \text{O} \\ \parallel \\ \text{HCOH} \end{array}$	<i>-benzoic acid</i> 
Ester	Name alkyl group on carboxylate O + add <i>-ate</i> to common prefix name $\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_3\text{COCH}_2\text{CH}_3 \end{array}$	Name alkyl group on carboxylate O + change <i>-e</i> to <i>-oate</i> in base name $\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_3\text{COCH}_2\text{CH}_3 \end{array}$	Name alkyl group on carboxylate O + add suffix <i>-benzoate</i> 

Acid Halide	Add <i>-yl</i> halide to common prefix name 	Change <i>-e</i> to <i>-oyl</i> halide in base name 	<i>-benzoyl halide</i> 
Amide	Add <i>-amide</i> (alkyl groups on N designated with <i>N</i> and named 1st) 	Change <i>-e</i> to <i>-amide</i> (alkyl groups on N designated with <i>N</i> and named 1st) 	<i>-benzamide</i> 
Nitrile	Add <i>-onitrile</i> to common prefix name 	Add <i>-nitrile</i> to base name (**carbon of <i>-CN</i> included when counting) 	<i>-benzonitrile</i> 
Numbering Substituents	Use Greek numbers starting from α -carbon 	Use numbers. Carbonyl carbon gets a number 	Number or use <i>o, m, p</i> 
Common Prefix Names	R in:  R = H "form" R = CH ₃ - "acet" R = CH ₃ CH ₂ - "propion" R = CH ₃ CH ₂ CH ₂ - "butyr"	R = CH ₃ CH ₂ CH ₂ CH ₂ - "valer" R = CH ₃ CH ₂ CH ₂ CH ₂ CH ₂ - "capro" R = CH ₃ (CH ₂) ₁₀ - "laur" R = CH ₃ (CH ₂) ₁₆ - "stear"	

Priority: acid > anhydride > ester > acid halide > amide > nitrile > aldehyde > ketone > alcohol > amine > ether